

The fabrication and magnetic properties of nanowire-like iron oxide

This article has been downloaded from IOPscience. Please scroll down to see the full text article.

2004 J. Phys.: Condens. Matter 16 4541

(<http://iopscience.iop.org/0953-8984/16/25/011>)

View [the table of contents for this issue](#), or go to the [journal homepage](#) for more

Download details:

IP Address: 129.252.86.83

The article was downloaded on 27/05/2010 at 15:37

Please note that [terms and conditions apply](#).

The fabrication and magnetic properties of nanowire-like iron oxide

L Y Zhang, D S Xue¹, X F Xu, A B Gui and C X Gao

Key Laboratory for Magnetism and Magnetic Materials of MOE, Lanzhou University,
Lanzhou 730000, People's Republic of China

E-mail: xueds@lzu.edu.cn

Received 2 April 2004

Published 11 June 2004

Online at stacks.iop.org/JPhysCM/16/4541

doi:10.1088/0953-8984/16/25/011

Abstract

Nanowire-like iron oxides (α -Fe₂O₃, Fe₃O₄ and γ -Fe₂O₃) of diameter about 200 nm and length up to 8 μ m were assembled into anodic aluminium oxide templates through electrodeposition followed by a heat-treating process. These wires have polycrystalline structures and consist of fine particles whose magnetic moments have preferred orientations in the templates. Some unique magnetic properties, such as perpendicular magnetic anisotropy in Fe₃O₄ wires, reduced transition temperature in α -Fe₂O₃ and Fe₃O₄ wires, and superparamagnetism in γ -Fe₂O₃ wires, were observed. The properties can be understood by their unique structures induced by reduced dimension and the competition of the shape and crystalline anisotropy.

1. Introduction

Ordered quasi-one-dimensional (1D) magnetic nanostructures have been of great interest both for theoretical study and for potential nanodevice applications due to their peculiar structural characteristics, size effects and some novel physical properties that are different from the bulk materials [1, 2]. A large variety of quasi-1D nanostructures, such as nanotubes, nanorods, nanowires, and nanobelts, have been synthesized by various techniques. One of the commonly used methods for generating quasi-1D nanostructures is based on the electrodeposition of various materials in porous anodic aluminium oxide (AAO) templates [3, 4]. This technique has been applied to fabricate metal and alloy nanowires as parallel arrays that displayed unique magnetic properties (e.g. anisotropy magnetization) [5, 6]. It is, however, limited for use with oxide-based magnetic materials (e.g. ferrite), mainly due to technical difficulties involved in the formation of oxides through electrochemical deposition. Developing a new and, in particular, cost-effective technique for fabricating quasi-1D ferrite nanostructures is of great scientific

¹ Author to whom any correspondence should be addressed.

and commercial interest. We have developed a low-cost method based on electrodepositing a precursor β -FeOOH into the AAO templates, followed by a heat-treating process. In earlier work, we reported the fabrication and some properties of α -Fe₂O₃ [7] and Fe₃O₄ [8] wires. γ -Fe₂O₃ wires were also synthesized recently. In our extensive research, we found that these wires are composed of fine particles and the magnetic moments of these particles have preferred orientations. Also their magnetic properties can be interpreted satisfactorily by these unique structures. We think that these nanostructures might more suitably be named nanowire-like systems. In this paper, we report the fabrication process and magnetic properties of γ -Fe₂O₃ wires. We also summarize and compare the properties of α -Fe₂O₃, Fe₃O₄ and γ -Fe₂O₃ wires.

2. Experiment

The detailed fabrication procedure of AAO templates is described elsewhere [9]. In this study, a two-step anodization process was employed [10] and all the templates were anodized at 120 V_{DC} for 3 h in 0.5 M phosphoric acid. After the anodization, the AAO templates were electrodeposited for 10 min in a mixed solution containing 9 g l⁻¹ FeCl₃·6H₂O and 14 g l⁻¹ (NH₄)₂C₂O₄·H₂O, where the voltage and frequency are 15 V_{AC} and 70 Hz, respectively. Then precursor β -FeOOH nanowires can be obtained [11].

After removing the remaining aluminium layers in a saturated HgCl₂ solution, the precursor membranes were placed in a horizontal tube furnace and calcined in air at 600 °C for 2 h. The product was characterized as α -Fe₂O₃ [7]. Then the α -Fe₂O₃ wires were reduced at 325 °C for 2 h in a hydrogen flow of 30 ml min⁻¹, and the product was verified as Fe₃O₄ [8]. γ -Fe₂O₃ wires can be obtained when Fe₃O₄ wires are heat-treated at 300 °C for 2 h in air.

3. Results and discussion

The detailed report on the preparation and characterization of the precursor β -FeOOH can be found in [11]. The following are related to the characterization and properties of three iron oxides nanostructures after heat-treating the precursor wires.

Figure 1 shows the x-ray diffraction (XRD, Philips X'Pert) patterns of the three oxides with AAO templates. All of them have polycrystalline structures with broadening peaks. α -Fe₂O₃ wires corresponding to the standard JCPDS 33-0664 can be easily identified. However, the other two samples have almost the same diffraction peaks due to the same spinel structure for Fe₃O₄ and γ -Fe₂O₃. Mössbauer spectrum (MS) analysis was used to verify the phase further. The polycrystalline structures of these samples indicate that the crystal orientations are imperfect, which is different from most metal and alloy nanowires electrodeposited in AAO templates directly [5, 12]. It can be conjectured that our samples are composed of fine particles with different crystal orientations due to dehydration of the precursor and the ac current electrodepositing process.

Figure 2(a) shows a scanning electron microscope (JEO 2000 SEM) image of the top view of the templates. The average pore diameter is about 200 nm. The morphologies of the three oxides after dissolving the templates in 0.1 M NaOH solution are shown in figures 2(b)–(d), respectively. The insets display the corresponding selected area electron diffraction (SAED) patterns. The average diameter of the wires is about 200 nm, which is in good agreement with the size of the pores of the AAO templates, and the average length is up to 8 μ m. Since Fe₃O₄ and γ -Fe₂O₃ have similar SAED patterns, the inset pattern shown in figure 2(c) was taken on several wires, whereas the pattern in figure 2(d) was taken on an individual wire. The bright diffraction rings show the polycrystalline spinel structures which are in good agreement with

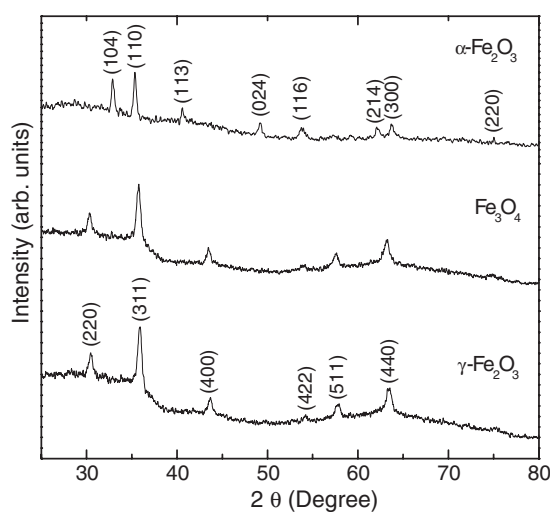


Figure 1. XRD patterns of three iron oxides (α - Fe_2O_3 , Fe_3O_4 and γ - Fe_2O_3) wires.

the XRD results. The irregularly arranged spots on the individual wire indicate the existence of fine crystallized particles with different crystal orientations.

MS of the three oxides in AAO templates with the γ -ray parallel to the wire axis are shown in figure 3. Since all MS at 300 K contain a large percentage of doublets, low temperature MS were collected to test that the doublets resulted from the contribution of superparamagnetic particles. Figure 3(a) shows the MS of α - Fe_2O_3 wires at 300 and 80 K. The 300 K MS is composed of a doublet and a sextet whose hyperfine field (H_{hf}) is about 50.11 T. Figures 3(b) and (c) show the MS of Fe_3O_4 and γ - Fe_2O_3 wires at 300 and 20 K. The 300 K MS in figure 3(b) consists of one doublet and two sextets in which the H_{hf} are 47.4 and 43.9 T, respectively. However, the 300 K MS in figure 3(c) is composed of one doublet and one sextet in which the H_{hf} is 46.5 T. From the different values of the H_{hf} , two oxides can be identified because the values of the hyperfine fields of the A- and B-site for Fe_3O_4 have an obvious difference; however, for γ - Fe_2O_3 the values are almost the same [13]. The H_{hf} of these wires are less than those of the bulk materials mainly due to the existence of collective magnetic excitation caused by the size distribution of the crystallites [14]. The reduced relative area of the doublet in the 80 K MS in figure 3(a) and the vanished doublets in the 20 K MS shown in figures 3(b) and (c) indicate that the doublets result from the contributions of the superparamagnetic phase.

The relative intensity ratio of the sextet for the three oxides at low temperature is different, indicating that the direction of the magnetic moments of the particles is different. The intensity ratio for the α - Fe_2O_3 wires in the 80 K MS is nearly 3:4:1, suggesting that the spins align almost perpendicular to the wire axis. However, for Fe_3O_4 and γ - Fe_2O_3 wires they are neither parallel nor perpendicular to the wire axis. This is different from the MS of metal and alloy nanowires in which lines 2 and 5 are almost absent, suggesting that the moments are perfectly aligned [12]. Moreover, for a given oxide, the ratios at room and low temperature are also different. This can be understood by the effect of the thermal agitation. The MS of the three oxides wires are also different from the isotropic particle systems where the moments aligned randomly. All of these differences might be related to the competition of the shape and the crystalline anisotropy.

The magnetic properties of these iron oxides wires at room temperature were investigated with a Lake Shore 7304 vibrating sample magnetometer. No remanence was observed in

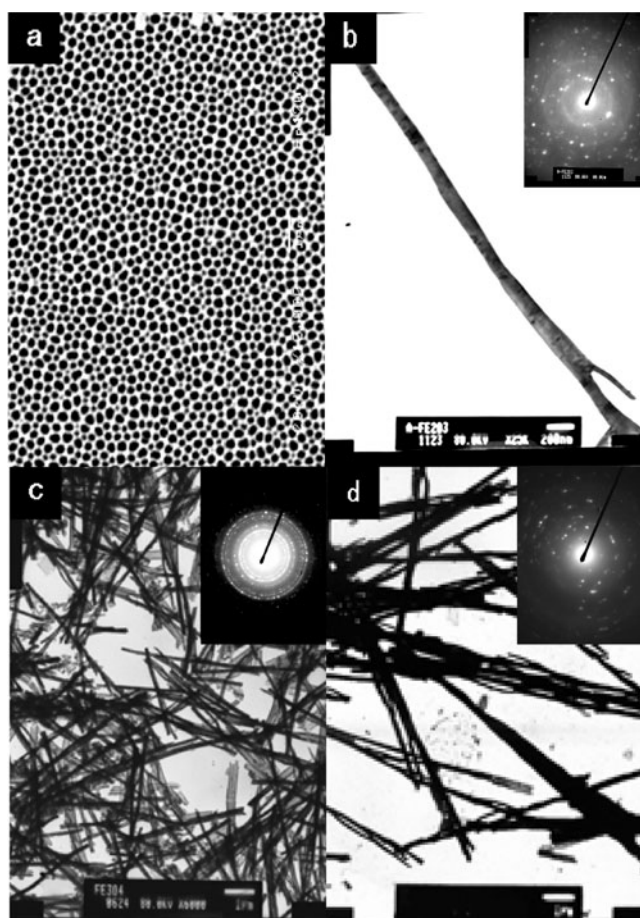


Figure 2. (a) SEM micrograph of the top view of AAO templates. (b)–(d) TEM images of α -Fe₂O₃, Fe₃O₄ and γ -Fe₂O₃ wires, respectively. The insets are the corresponding SAED patterns. The inset pattern in figure 2(c) was taken on several wires and in figure 2(d) was taken on an individual wire.

α -Fe₂O₃ wires at room temperature. However, when the temperature decreased to 80 K with an applied field of 5 T, weak ferromagnetism (WF) was observed (shown in figure 4(a) measured on a quantum design Magnetic Property Measurement System). The magnetic moments in WF state may have contributions from both the canting of the sublattice magnetization directions and from uncompensated spins as observed for α -Fe₂O₃ nanoparticles [15], because our wires are composed of fine α -Fe₂O₃ particles. Figure 4(b) shows the hysteresis loops of Fe₃O₄ wires at room temperature, in which the H_{\parallel} and H_{\perp} indicate that the external field was applied parallel and perpendicular to the wire axis, respectively. The expected magnetic anisotropy was found. The coercivity (H_c) and remanence ratio (M_r/M_s) are 378 Oe and 0.6 in the parallel direction, which are higher than 147 Oe and 0.1 in the perpendicular one. Furthermore, in the parallel direction, technical saturation occurs in fields of about 2 kOe, whereas in the perpendicular direction the wires are not magnetically saturated in a field of 8 kOe. This suggests that the magnetic moments of Fe₃O₄ particles in the templates have preferred orientations due to the shape anisotropy. However, from MS results they are not parallel to the wire axis. Figure 4(c) shows the hysteresis loops for the γ -Fe₂O₃ wires at room temperature which

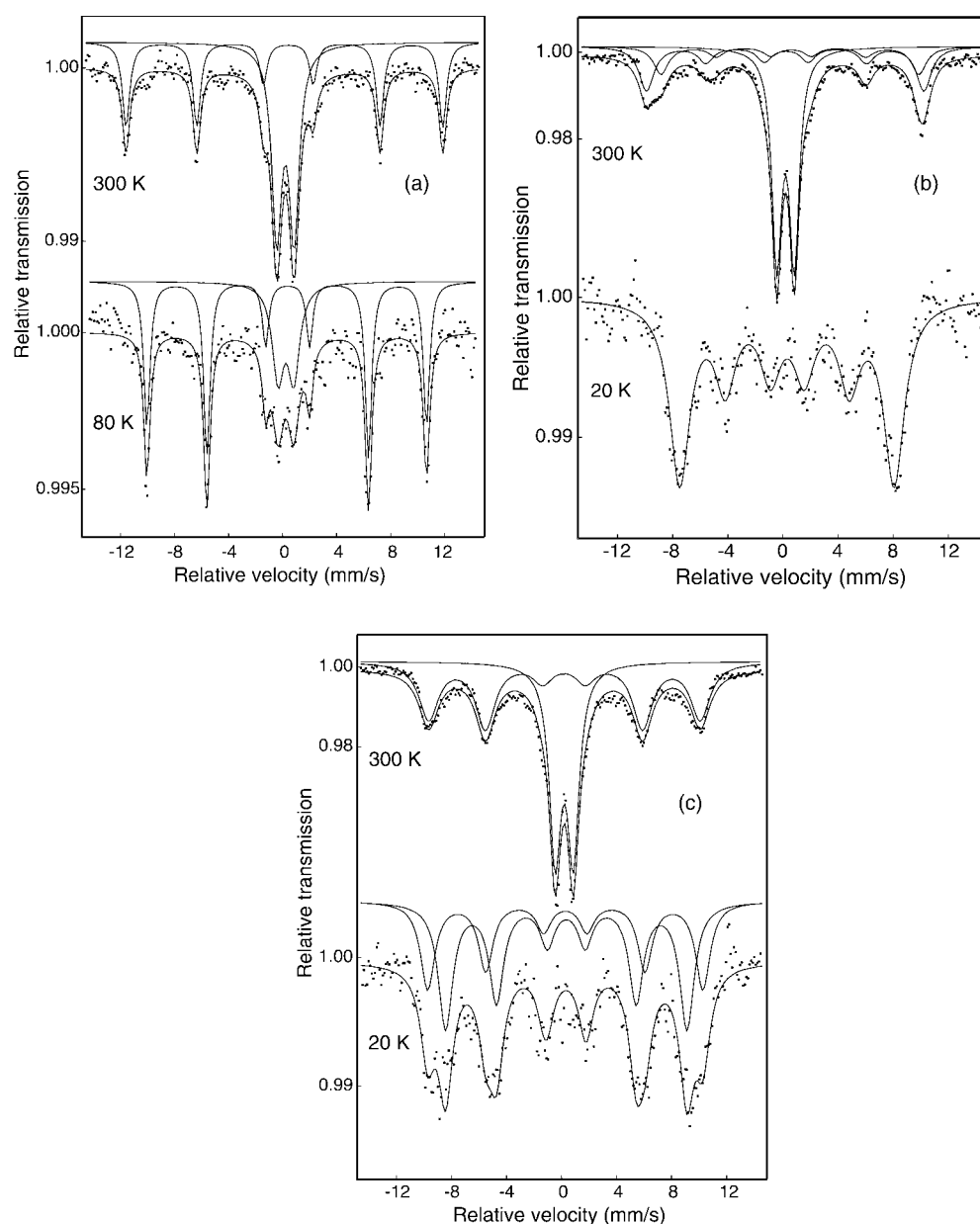


Figure 3. MS of (a) α - Fe_2O_3 wires at 300 and 80 K, (b) Fe_3O_4 wires at 300 and 20 K, and (c) γ - Fe_2O_3 wires at 300 and 20 K.

are close to superparamagnetic behaviour. This should be attributed to the size of γ - Fe_2O_3 particles that are in the superparamagnetic scale, and the magnetic moments of the particles cannot be stabilized due to thermal activation. The different saturated magnetic moments in the parallel and perpendicular directions are due to the different demagnetization fields in the two directions.

Figure 5 shows the temperature dependence of the magnetization for α - Fe_2O_3 and Fe_3O_4 wires, respectively. For α - Fe_2O_3 , below 80 K the magnetization increases with increasing

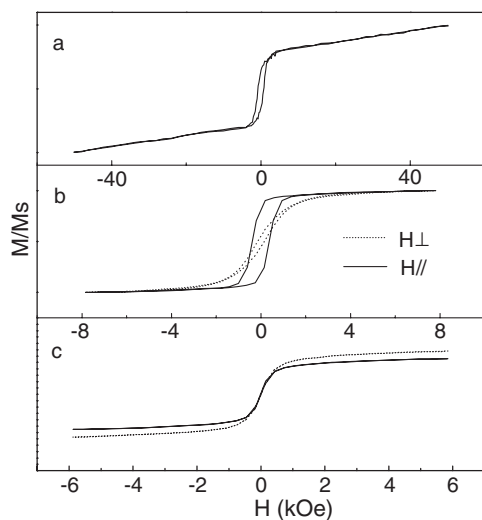


Figure 4. Hysteresis loops of (a) α - Fe_2O_3 wires at 80 K, (b) Fe_3O_4 wires at room temperature, and (c) γ - Fe_2O_3 wires at room temperature.

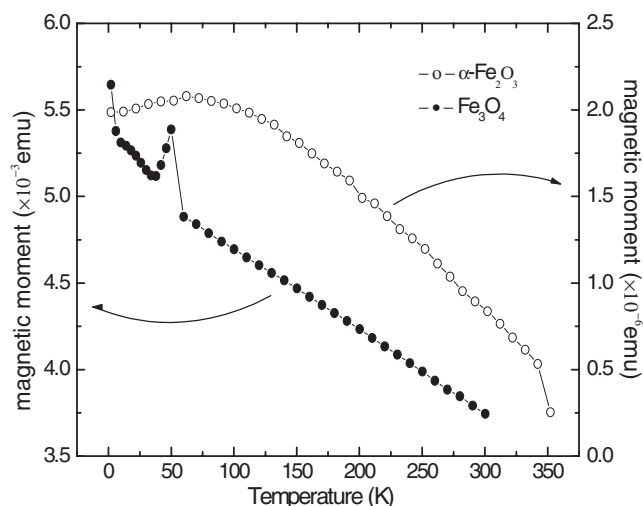


Figure 5. Temperature dependence of the magnetization for α - Fe_2O_3 and Fe_3O_4 wires, respectively.

temperature, which is characteristic of antiferromagnetism. Above 80 K it decreases with increasing temperature. This behaviour is more like that of ferromagnetism which corresponds well with the WF of α - Fe_2O_3 above the Morin transition temperature (T_M). When the temperature is at about 350 K the magnetization sharply decreases to near zero. This temperature can be explained as the Néel transition temperature (T_N). So, for α - Fe_2O_3 wires the T_M is about 80 K and the T_N is about 350 K, which are much less than those of the bulk materials. A decreased T_M value was also observed on small α - Fe_2O_3 particles [16].

For Fe_3O_4 wires measured at 2 T the magnetization increases as the temperature decreases. When the temperature decreases to 50 K the magnetization first decreases and then increases with decreasing temperature. The change of the magnetization at 50 K can be interpreted

by a Verwey transition where the crystal structure changes from cubic to monoclinic [17]. The transition temperature (T_V) is much lower than that of the bulk materials of about 120 K, which is due to the reduced dimension. A reduced T_V value was also observed on nanosized particles [18] and films [19]. The decreased magnetization at T_V was explained by Muxworthy [20] as the formation of an e-sublattice in which the e-sublattice electrons align antiparallel to the net magnetization of the A- and B-sublattice when cooling through T_V . He attributed the increased magnetization below T_V to the paraprocess in which the e-sublattice was destroyed or aligned in the direction of the field when the applied field is large enough to cause technical saturation. Our experimental results seem to be interpreted by this model. However, the jump of the saturation magnetization at T_V is not much sharper than that observed by Muxworthy. This can be attributed to the particle size distribution in the wires. The variations in stoichiometry can also strongly affect T_V [21]. In addition, it must be mentioned that the sharply increased magnetization at very low temperature might have a contribution from the superparamagnetism.

4. Conclusion

Nanowire-like iron oxide (α -Fe₂O₃, Fe₃O₄ and γ -Fe₂O₃) wires of diameter about 200 nm and length up to 8 μ m were successfully synthesized in AAO templates. The wires have polycrystalline structures and consist of fine particles. The magnetic moments of these particles have preferred orientations. The moments of the α -Fe₂O₃ particles in the templates are almost perpendicular to the wire axis, whereas they are neither parallel nor perpendicular to the axis for Fe₃O₄ and γ -Fe₂O₃ particles. At room temperature, perpendicular magnetic anisotropy is observed clearly only in Fe₃O₄ wires, and no remanence was found in α -Fe₂O₃ and γ -Fe₂O₃ wires. The transition temperatures T_M and T_N for α -Fe₂O₃ wires and T_V for Fe₃O₄ wires are reduced to about 80, 350 and 50 K, respectively, which are much less than the values for bulk materials. These unique magnetic properties result from the reduced dimensions as well as the induced preferred orientations of the magnetic moments of the oxide particles in the wires.

Acknowledgments

This work is supported by the NSFC (Grant Nos 10374038 and 50171032), the Trans-Century Training Program Foundation for the Talent, and the Excellent Young Teachers Program of MOE, People's Republic of China.

References

- [1] Alivisatos A P 1996 *Science* **271** 933
- [2] Gulseren O, Ercolessi F and Tosatti E 1998 *Phys. Rev. Lett.* **80** 3775
- [3] Zeng H, Zheng M, Skomski R, Sellmyer D J, Liu Y, Menon L and Bandyopadhyay S 2000 *J. Appl. Phys.* **87** 4718
- [4] Metzger R M, Konovalov V V, Sun M, Xu T, Zangari G, Xu B, Benakli M and Doyle W D 2000 *IEEE Trans. Magn.* **36** 30
- [5] Zeng H, Skomski R, Menon L, Liu Y, Bandyopadhyay S and Sellmyer D J 2002 *Phys. Rev. B* **65** 134426
- [6] Pang Y T, Meng G W, Shan W J, Fang Q and Zhang L D 2003 *Chin. Phys. Lett.* **20** 144
- [7] Xue D S, Gao C X, Liu Q F and Zhang L Y 2003 *J. Phys.: Condens. Matter* **15** 1455
- [8] Xue D S, Zhang L Y, Gao C X, Xu X F and Gui A B 2004 *Chin. Phys. Lett.* **21** 733
- [9] Masuda H and Fukuda K 1995 *Science* **268** 1466
- [10] Li A P, Müller F, Birner A, Nielsch K and Gösele U 1998 *J. Appl. Phys.* **84** 6023
- [11] Gao C X, Liu Q F and Xue D S 2003 *J. Mater. Sci. Lett.* **21** 1781
- [12] Chen Z Y, Zhan Q F, Xue D S, Li F S, Zhou X Z, Kunkel H and Williams G 2002 *J. Phys.: Condens. Matter* **14** 613
- [13] Deniels J M and Rosencwaig A 1969 *J. Phys. Chem. Solids* **30** 1561

-
- [14] Borzi R A, Stewart S J, Punte G, Mercader R C, Mansilla M V, Zysler R D and Cabanillas E D 1999 *J. Magn. Mater.* **205** 234
- [15] Néel L 1962 *Low Temperature Physics* ed C DeWitt, B Dreyfus and P G DeGennes (London: Gordon and Breach) p 411
- [16] Yamamoto N 1968 *J. Phys. Soc. Japan* **24** 23
- [17] Yoshida J and Iida S 1979 *J. Phys. Soc. Japan* **47** 1627
- [18] Goya G F, Berquo T S and Fonseca F C 2003 *J. Appl. Phys.* **94** 3520
- [19] Margulies D T, Parker F T, Spada F E, Goldman R S, Li J, Sinclair R and Berkowitz A E 1996 *Phys. Rev. B* **53** 9175
- [20] Muxworthy A R and McClelland E 2000 *Geophys. J. Int.* **140** 101
- [21] Honig J M 1995 *J. Alloys Compounds* **229** 24